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Exhibit 2-4 Use the convention and the crossed arrow to show the direction of the expected polarity of the indicated bonds in the following compounds.

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Exchange the positions
of the bond and the
electrons in the p
orbital to draw the
resonance form of each
grouping. (a) Methyl
phosphate anion has 3
three-atom groupings
and thus has 3

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resonance forms. 9. 28

Chapter 2 (b) (c) (d)

2.11 When an acid
loses a proton, the

product is the
conjugate base of the
acid.

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